# CBCS Scheme

|     |  | <br> |  |  |  | p. Mar. |        |
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| USN |  |      |  |  |  |         | 15NT35 |
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# Third Semester B.E. Degree Examination, Dec.2017/Jan.2018 Physical and Chemical Principles of Nano Technology

Time: 3 hrs. Max. Marks: 80

Note: Answer any FIVE full questions, choosing ONE full question from each module.

# Module-1

| 1 | a. | Define quantum mechanics. S | State uncertainty | principle with an | example and add a note on |
|---|----|-----------------------------|-------------------|-------------------|---------------------------|
|   |    | complementarity.            |                   |                   | (05 Marks)                |

- b. Give a brief note on the Molecular Orbital (MO) theory and its applications. (08 Marks)
- c. Write a short note on computational chemistry and name few applications. (03 Marks)

### OR

- 2 a. State and explain dual nature of matter by Debroglie.
  b. Give brief note on the valence bond theory and its applications.
  (03 Marks)
  (08 Marks)
  - c. Enumerate the Planck's hypothesis.

### Module-2

- 3 a. State and explain the first law of thermodynamics with mathematical expressions. (06 Marks)
  - b. What is spontaneous process? Write the criteria for spontaneity. (04 Marks)
  - c. What is third law of thermodynamics? Write the illustration. (06 Marks)

### OR

- 4 a. Explain the concept of heat capacity at constant volume and constant pressure and obtain the relation between CV and CP. (06 Marks)
  - b. Define and explain the following thermodynamic terms with example:
    - i) System and surroundings
    - ii) Internal energy
    - iii) Enthalpy. (06 Marks)
  - c. State the laws of thermodynamics with necessary equations.

# Module-3

- 5 a. Explain Debye theory of molar heat capacity and limitations. (08 Marks)
  - b. Derive Kronig Penny model with necessary equations and graphs.

## (08 Marks)

(04 Marks)

(05 Marks)

### OR

- 6 a. Distinguish between metals, insulators and semiconductors. (06 Marks)
  - b. Describe about Brillouin zones. (04 Marks)
  - c. Explain classical theory of molar heat capacity and limitations. (06 Marks)

1 of 2

(06 Marks)

# Module-4

| b. Discuss the concept of tunneling. (04   | Marks)<br>Marks)<br>Marks) |
|--|----------------------------|
| <ul><li>ii) Extrinsic semiconductors.</li><li>b. Discuss the concept of tunneling.</li><li>(04)</li></ul>  | Marks)<br>Marks)           |
| b. Discuss the concept of tunneling. (04   | Marks)<br>Marks)           |
|  | Marks)                     |
| o. Brief about emassion (vs) quantum tumoring.   |                            |
|  | Marks)                     |
|  | Marks)                     |
| OR (Section 1)   | Marks)                     |
|  | mains)                     |
|  | Maulta                     |
|  | Marks)                     |
| c. Write about Donor and acceptor levels in semiconductors. (06  | Marks)                     |
|  |                            |
|  |                            |
| The state of the s |                            |
| Module-5   |                            |
|  | Marks)                     |
| b. Describe about the following electric properties of colloids:   |                            |
|  | Marks)                     |
| c. Explain about the characteristics and identification of types of emulsions. (05)  | Marks)                     |
|  |                            |
|  |                            |
| OR   |                            |
| 10 a. What are colloids? Explain about crystalloids and colloids. (04)   | Marks)                     |
| b. Discuss the following optical properties of colloids:   |                            |
| i) Visibility ii) colour iii) tyndall effect. (06  | Marks)                     |
| c. Describe about types of emulsions formed on mixing of two partly or completely ins  | soluble                    |

\* \* \* \*

liquids.